

Chemical Science International Journal

Volume 33, Issue 6, Page 155-162, 2024; Article no.CSIJ.127339 ISSN: 2456-706X (Past name: American Chemical Science Journal, Past ISSN: 2249-0205)

Development of ZrO2-Coated SPME Fibers for Detecting Benzotriazole Ultraviolet Filters in Water Samples

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Authors' contributions

This work was carried out in collaboration between both authors. Both authors read and approved the final manuscript.

Article Information

DOI[: https://doi.org/10.9734/CSJI/2024/v33i6934](https://doi.org/10.9734/CSJI/2024/v33i6934)

Open Peer Review History:

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Original Research Article

Received: 26/09/2024 Accepted: 29/11/2024 Published: 02/12/2024

ABSTRACT

In this study, the zirconium dioxide nanoparticles $(ZrO₂NPs)$ were coated on the surface of stainless steel (SS) by cyclic voltammetry (CV) as solid phase microextraction fiber coating for detection of the benzotriazole ultraviolet filters (BUvFs) in different real water samples. The composition and images of the new SS@ZrO₂NPs fiber were characterized by scanning electron microscopy (SEM). The extraction capability of SS@ZrO₂NPs fiber was investigated for the enrichment and detection of benzotriazole ultraviolet filters (BUvFs) phthalate acid esters (PAEs) and polycyclic aromatic hydrocarbons (PAHs) coupled with HPLC. The fabricated SS@ZrO₂NPs fiber showed excellent extraction selectivity and good extraction capability for BUvFs. Under the optimized conditions, the SS@ZrO2NPs-SPME method presented linear ranging from 0.15 to 200 µg/L with correlation coefficients of higher than 0.9982. Moreover the limits of detection (LODs) varied from 0.020 µg/L to 0.138 µg/L. Relative standard deviations (RSDs) with single fiber were below 7.78% and 8.91% for intra-day and inter-day measurements, respectively. The developed method was used to the detection of trace BUvFs in different real water samples.

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Cite as: Shen, Xiaoyan, and Huiju Wang. 2024. "Development of ZrO2-Coated SPME Fibers for Detecting Benzotriazole Ultraviolet Filters in Water Samples". Chemical Science International Journal 33 (6):155-62. https://doi.org/10.9734/CSJI/2024/v33i6934.

Keywords: Zirconium dioxide; solid phase microextraction; stainless steel; benzotriazole ultraviolet filters; high-performance liquid chromatography.

1. INTRODUCTION

"Solid phase microextraction (SPME) is a very useful preparation technique. It integrate sampling, extraction, enrichment and direct sample introduction into one step" [1]. "As an effective sample preparation technology, it had been applied to many fields such as environmental food, pharmaceutical and biological analyses" [2-6]. "Some commercial fibers show some defects such as fragility, the stripping of coatings, bending of the needle and easily swelling in an organic solvent. These limit its wide applications" [7]. Therefore, in the past two decades, the development of new fibers has become a hot topic in the field [8]. Thus, a large number of new materials with good adsorption properties are constantly being developed as SPME fiber coatings [9,10].

"ZrO² has the advantages of durability, corrosion resistance, wear resistance, strong adhesion to metal surface, non-toxic and so on, and is one of the most promising inorganic materials" [11]. "Due to good sorptive properties, ZrO² nanomaterials with various structures have been successfully applied as fiber coating for SPME" [12,13].

"As a class of benzotriazole derivatives, BUvFs are widely used in industrial products and personal care products to protect against ultraviolet radiation from sunlight" [14]. Because of their overuse, BUvFs have been found in environmental substrates such as surface water, biomes, wastewater and sediments [14-16]. "Some BUvFs have been reported to have endocrine-disrupting and estrogenic effects in organisms" [17,18]. As a result, they have been recognized as emerging pollutants of concern, and their excessive use has been restricted in some countries and regions [18]. "However, BUvFs is present at trace or ultra-trace levels in environmental samples. Therefore, sample pretreatment is necessary for the analysis of trace analytes in complex substrates". In this study, the SS@ZrO₂NPs fiber was prepared and used to the extraction and determination of BUvFs in environmental water samples coupled with HPLC-UV.

2. METHODOLOGY

2.1 Reagents and Materials

Bare stainless steel was purchased from Gaoge. Zirconium (IV) oxychloride octahydrate (ZrOCl2• 8H₂O) was obtained from Huawei Ruike (Beijing, China). Sodium chloride (NaCl) and Potassium chloride (KCl) were obtained from Shuangshuang Chemical Ltd (Shandong, China). Hydrofluoric acid (HF) was purchased from Shanghai Hunter Fine Chemicals Ltd. (Shanghai, China). The methanol with HPLC-grade was supplied from Yuwang Chemical Company (Shandong, China). 2- (2H-benzotriazol-2-yl)-4- (1,1,3,3-tetramethylbutyl) phenol (UV-329), 2- (2′ hydroxy-5′-methylphenyl) benzotriazole (UV-P), 2-tertbutyl-6-(5-chloro-2H-benzotriazol-2-yl)-4 methylphenol (UV-326), 2-(3,5-di-tert-butyl-2 hydroxyphenyl)-2H-benzotriazole (UV-320) and 2-(2H-benzotriazol-2-yl)-4,6- bis(1-methyl-1 phenylethyl) phenol (UV-234) were from AccuStandard (New Haven, CT, USA). Fuoranthene (Flu), benzo[a] pyrene (B[a]p), naphthalene (Nap), phenanthrene (Phe)and, di- (2-ethylhexyl) phthalate (DEHP), di-n-butyl phthalate (DBP), dimethyl phthalate (DMP), di-noctyl phthalate (DOP) were provided from Sigma-Aldrich (St. Louis, MO, USA)

2.2 Apparatus

Images of the prepared fibers were examined using Ultra Plus microscopy (Zeiss, Oberkochen, Germany), together with energy dispersive X-ray spectroscopy (EDX) analysis. Separation and determination of analytes were performed on Agilent 1260 Infinity LC system (Waldbronn, Baden-Württemberg, Germany) equipped with variable wavelength detector and a SB-C18 column. The data was collected with an Agilent ChemStation. Ultrapure water was obtained from a XYG-60-H water purification system (Beijing, China).

2.3 Prepared of the SS@ZrO2NPs Fiber

The SS fibers were washed in an ultrasonic bath with acetone and ultra-pure water for 10 minutes, respectively. One end of the SS wire was pretreated in 40% (w/w) HF solution for 30 minutes and then rinsed in an ultrasonic water bath for 5 minutes. Thereafter ZrO₂ nanoparticles were deposited on the surface of SS by CV between 2.0 and 0 V (versus Ag/AgCl) at a scanning rate of 20 mV/s for 15 cycles. The electrolyte is composed of 0.005 mol/L ZrOCl² and 0.1 mol/L KCl. After deposition, the SS@ZrO2NPs fiber was gently washed with ultrapure water and dried in air.

2.4 Extraction Procedure

10 mL sample solution was added into a sample vial with a PEEK cap. The SS@ZrO2NPs fiber was usually immersed in a stirred sample solution and extracted at a selected temperature for a specified time. After extraction, the SS@ZrO₂NPs fiber was removed from the sample vial and introduced into the SPME-HPLC interface for static desorption in corresponding mobile phase. The mobile phase of methanol/water with 95/5 (v/v), 90/10 (v/v) and 85/15 (v/v) was used for HPLC analysis to BUvFs, PAHs, PAEs, respectively. The flow rate was 1 mL/min. Corresponding detection wavelength were set at 340 nm, 254 nm and 280 nm. Before extraction, the $SS@ZrO₂NPs$ fiber was washed with methanol for 5 minutes.

3. RESULTS AND DISCUSSION

3.1 Characterization of SS@ZrO2NPs Fiber

The surface images of SS@ZrO2NPs fiber was characterized by SEM. As can be seen in Fig. 1B, bare SS fiber exhibited smooth surface microscopically at high magnification. Moreover,

EDX analysis shows that the main surface components of SS fiber are Fe, Ni, Cr and Si elements. After deposition, the uniform and porous nanoparticles were formed on the SS supports. EDX analysis (Fig. 2B) exhibited that the strong characteristic peaks of Zr and O elements appear. This result revealed the SS@ZrO2NPs fiber was successfully fabricated.

3.2 Extraction Selectivity

Fig. 3 shows the extraction selectivity of the SS@ZrO2NPs fber, which is estimated by three typical organic compounds including PAEs, PAHs and BUvFs. The SS@ZrO2NPs fiber shows poor extraction capability for PAEs (Fig. 3(b)) and PAHs (Fig. 3(d)). However the SS@ZrO2NPs fiber displays good extraction capability for BUvFs (Fig. 3(f)). This result is attributed to the inherent chemical properties of the ZrO2NPs coating. "Due to hydrophobic interactions between ZrO₂NPs coating and aromatic compounds, the SS@ZrO2NPs fiber shows good extraction capability for BUvFs. Furthermore the strong Lewis acidic sites on the surface of ZrO₂ nanoparticles is expected to have a good affinity for target analytes with a Lewis basic character" [12].

Fig. 1. SEM images of SS fiber (A) Bare SS wire (×500); (B) Bare SS wire (×10,000); (C) SS@ZrO2NPs fiber (×500); (D) SS@ZrO2NPs fiber (×10,000)

Fig. 2. EDX spectra of SS fiber. (A) Bare SS fiber; (B) SS@ZrO2NPs fiber

Fig. 3. Chromatograms of direct HPLC (a, c, e) and SPME-HPLC with the SS@ZrO2NPs fiber (b, d, f). PAEs (a, b), PAHs (c, d) and BUvFs (e, f) at a spiking level of 50 µg/L for each analyte

3.3 Extraction Conditions

The main SPME conditions including extraction time, temperature, stirring rate and ionic strength were studied and optimized. As can be seen in Fig. 4, the SS@ZrO2NPs fiber showed excellent extraction capability for BUvFs at 30 $°C$ for 40 min with a stirring rate of 500 r/mim. Typically, adding NaCl to the sample solution can increases the ionic strength and reduces the solubility of BUvFs in the aqueous phase. Herein, NaCl concentration ranging from 0 to 25% (w/v) was added and the extraction efficiency of SS@ZrO2NPs fiber for target BUvFs was investigated. As shown in Fig. 4(d), 5% (w/v) NaCl is added in the sample solution the optimal extraction capability of the novel fiber for BUvFs is reached. Therefore, 5% (w/v) NaCl is chosen as suitable salt

concentration. In addition, 5 min was employed as desorption time.

3.4 Method Validation

Under the optimized conditions, the analytical performance of the SPME-HPLC method with the SS@ZrO2NPs fiber was investigated for BUvFs. The results are listed in Table 1. In the range of $0.15 \sim 200$ µg/L, the linearity is good, and the correlation coefficient (r^2) is greater than 0.9982. Limits of detection (LODs) calculated with a signal-to-noise ratio of 3 (S/N=3) ranges from 0.020 µg/L to 0.138 µg/L. Relative standard deviations (RSDs) for BUvFs were less than 8.91 % with a single fiber at the spiking level of 50 µg/L. These data indicated show that the SPME-HPLC-SS@ZrO2NPs method has good accuracy, precision and sensitivity.

Table 2. Analytical results of BUvFs in different real water samples (n=3)

Fig. 4. Main SPME conditions. Extraction time (A); Temperature (B); Stirring rate (C) and Ionic strength (D)

3.5 Detection of Real Water Sample

In order to verify the feasibility of the
method, the prepared optical fiber is method, the prepared optical fiber is applied to the detection of target BUvFs in different water samples. All result were list in Table 2. To assess the accuracy of the results, standard BUvFs was added to the actual sample at levels of 15 µg/L and 30 µg/L, respectively. The desired recoveries ranging from 91.7% to 100.0% was obtained. And the RSDs was less than 8.4%. These results show that the new $SS@ZrO₂NPs$ fiber can be used for adsorption and sensitive detection of BUvFs in water samples in different environments.

4. CONCLUSIONS

The SS@ZrO2NPs fiber was successfully fabricated by cyclic voltammetry. Moreover the ZrO2NPs caoting possess porous and uniform structures. This special nanostructures facilitated the adsorption and extraction of analytes from

the solution into the fiber coating. The SS@ZrO2NPs fiber showed desired extraction capability for BUvFs. In addition, the SS@ZrO2NPs fiber was successfully applied for the detection of BUvFs in real water samples.

DISCLAIMER (ARTIFICIAL INTELLIGENCE)

Author(s) hereby declare that NO generative AI technologies such as Large Language Models (ChatGPT, COPILOT, etc) and text-to-image generators have been used during writing or editing of this manuscript.

ACKNOWLEDGEMENTS

The authors gratefully acknowledge financial support from the No.12M2024015.

COMPETING INTERESTS

Authors have declared that no competing interests exist.

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